

# Hybrid Organic Magnetic Metal Interfaces

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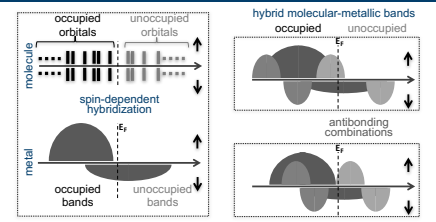
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## Introduction

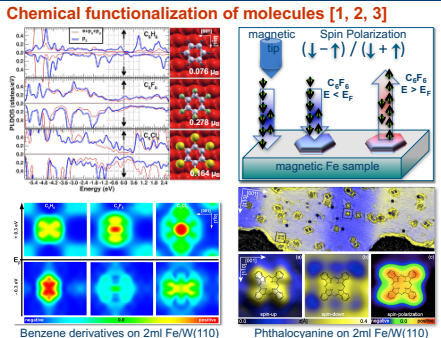
The density functional theory provides a framework with predictive power that can be used to describe organic-metal hybrid systems in a realistic manner. In this respect, *ab initio* studies elucidate how the subtle interplay between the electrostatic, the weak van der Waals and the strong chemical interactions determines the geometric, electronic and magnetic structure of hybrid organic-metal interfaces. More precisely, the interaction between the  $\pi$ -like electronic cloud of organic materials with the magnetic states of a metal substrate influences the (i) spin-polarization, (ii) magnetic exchange coupling, (iii) magnetic moments and (iv) their orientation at the hybrid interfaces.

This poster briefly summarizes how first-principles calculations (i) provide the basic insights needed to interpret surface-science experiments and (ii) are a key tool to design novel materials with tailored properties that can be integrated in carbon-based spintronic devices.

## Chemisorption: spin-dependent hybridization



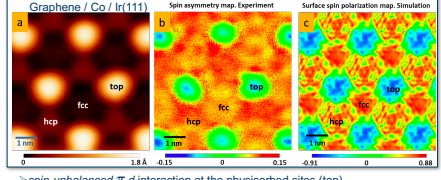
## Spin-polarization at the interface



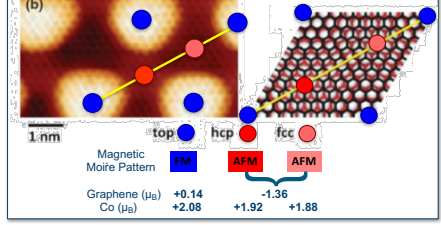
Benzene derivatives on 2ml FeW(110)      Phthalocyanine on 2ml FeW(110)

Magnetic spin-polarized current:  
 > conventional ferromagnetic surface: apply magnetic field  
 > molecule / ferromagnetic substrate: change bias voltage

## Moiré spin-unbalanced electronic structure [4]



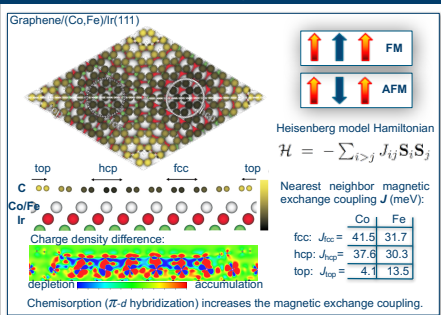
> spin-unbalanced  $\pi$ -d interaction at the physisorbed sites (top)  
 > inversion of the spin-polarization: ONLY at the chemisorbed sites (hcp and fcc)



Graphene Moiré Pattern

Magnetic	FM	AFM	AFM
Graphene ( $\mu_B$ )	+0.14	-1.36	-1.36
Co ( $\mu_B$ )	+2.08	+1.92	+1.88

## Magnetic exchange interactions



Nearest neighbor magnetic exchange coupling  $J$  (meV):

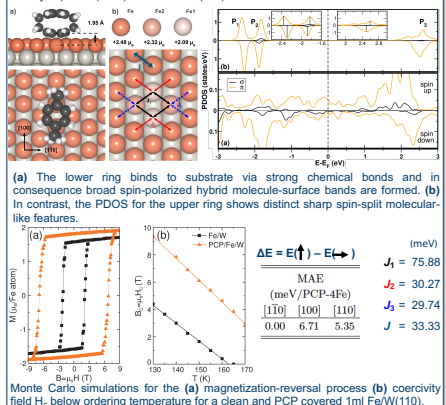
	Co	Fe
fcc: $J_{fcc}$	41.5	31.7
hcp: $J_{hcp}$	37.6	30.3
top: $J_{top}$	4.1	13.5

Charge density difference:  
 depletion      accumulation

Chemisorption ( $\pi$ -d hybridization) increases the magnetic exchange coupling.

## Magnetic hardening & softening with organics

### $\pi$ -d hybridization: intra-layer magnetic hardening [5]



(a) The lower ring binds to substrate via strong chemical bonds and in consequence broad spin-polarized hybrid molecule-surface bands are formed. (b) In contrast, the PDOS for the upper ring shows distinct sharp spin-split molecular-like features.

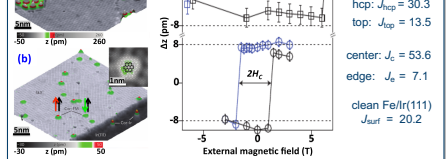
Monte Carlo simulations for the (a) magnetization-reversal process (b) coercivity field  $H_c$  under ordering temperature for a clean and PCP covered 1ml FeW(110).

$\Delta E = E(\uparrow) - E(\downarrow)$  (meV)

$J_1$	= 75.88
$J_2$	= 30.27
$J_3$	= 29.74
$J$	= 33.33

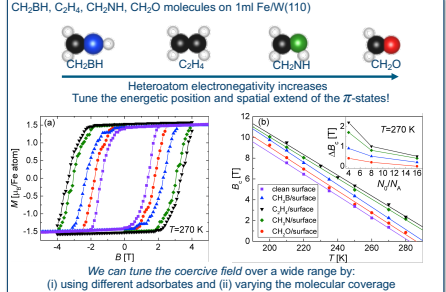
MAE (meV/PCP-4Fe):  
 [110] 0.00    [100] 6.71    [110] 5.35

### Atomic-like view of the magnetic hardening effect [6]



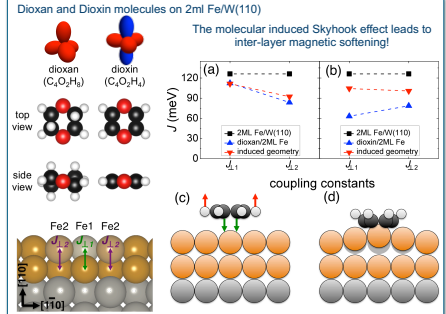
Magnetic exchange coupling  $J$  (meV):  
 fcc:  $J_{fcc} = 31.7$   
 hcp:  $J_{hcp} = 30.3$   
 top:  $J_{top} = 13.5$   
 center:  $J_c = 53.6$   
 edge:  $J_e = 7.1$   
 clean Fe/Ir(111)  $J_{surf} = 20.2$

### Chemical flexibility & magnetic hardening [7]

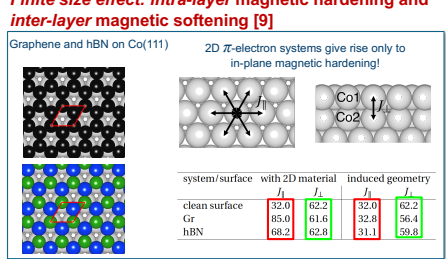


We can tune the coercive field over a wide range by:  
 (i) using different adsorbates and (ii) varying the molecular coverage

### Intra-layer magnetic softening with organics [8]



### Finite size effect: intra-layer magnetic hardening and inter-layer magnetic softening [9]

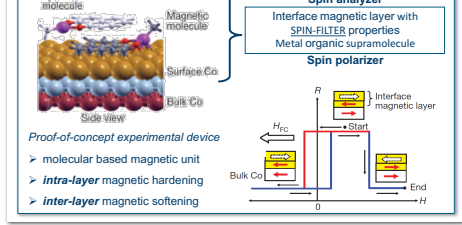


2D  $\pi$ -electron systems give rise only to in-plane magnetic hardening!

system/surface	with 2D material	induced geometry	
clean surface	32.0	62.2	32.0
Gr	85.0	61.6	32.8
hBN	68.2	62.8	31.1

## A nanomagnet embedded within a magnet

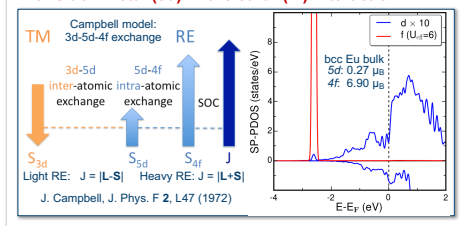
### Interface magnetoresistance effect [10]



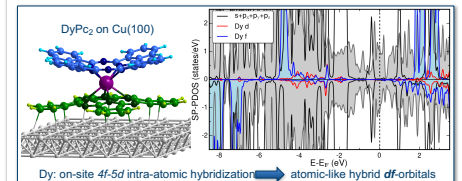
Proof-of-concept experimental device  
 > molecular based magnetic unit  
 > intra-layer magnetic hardening  
 > inter-layer magnetic softening

## Present and future investigations

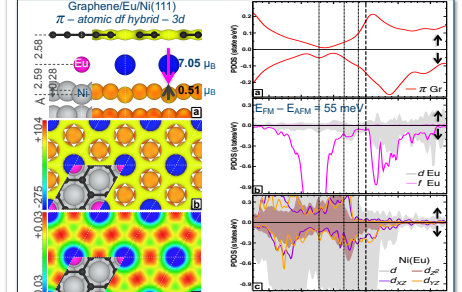
### Transition Metal (3d) – rare earth (4f) interaction



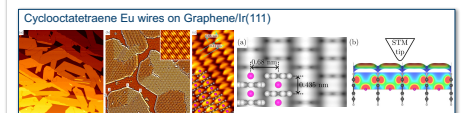
### On-site rare earth 4f – 5d hybridization [11, 12, 13]



Dy: on-site 4f-5d intra-atomic hybridization → atomic-like hybrid df-orbitals



### Cyclooctatetraene Eu wires on Graphene/Ir(111)



## Acknowledgments

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