

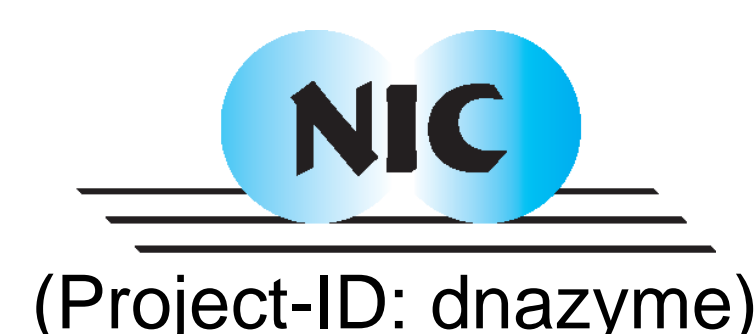
# DNAmazing: RNA cleavage mechanism of catalytically active DNA and the effect of non-natural nucleobases on it



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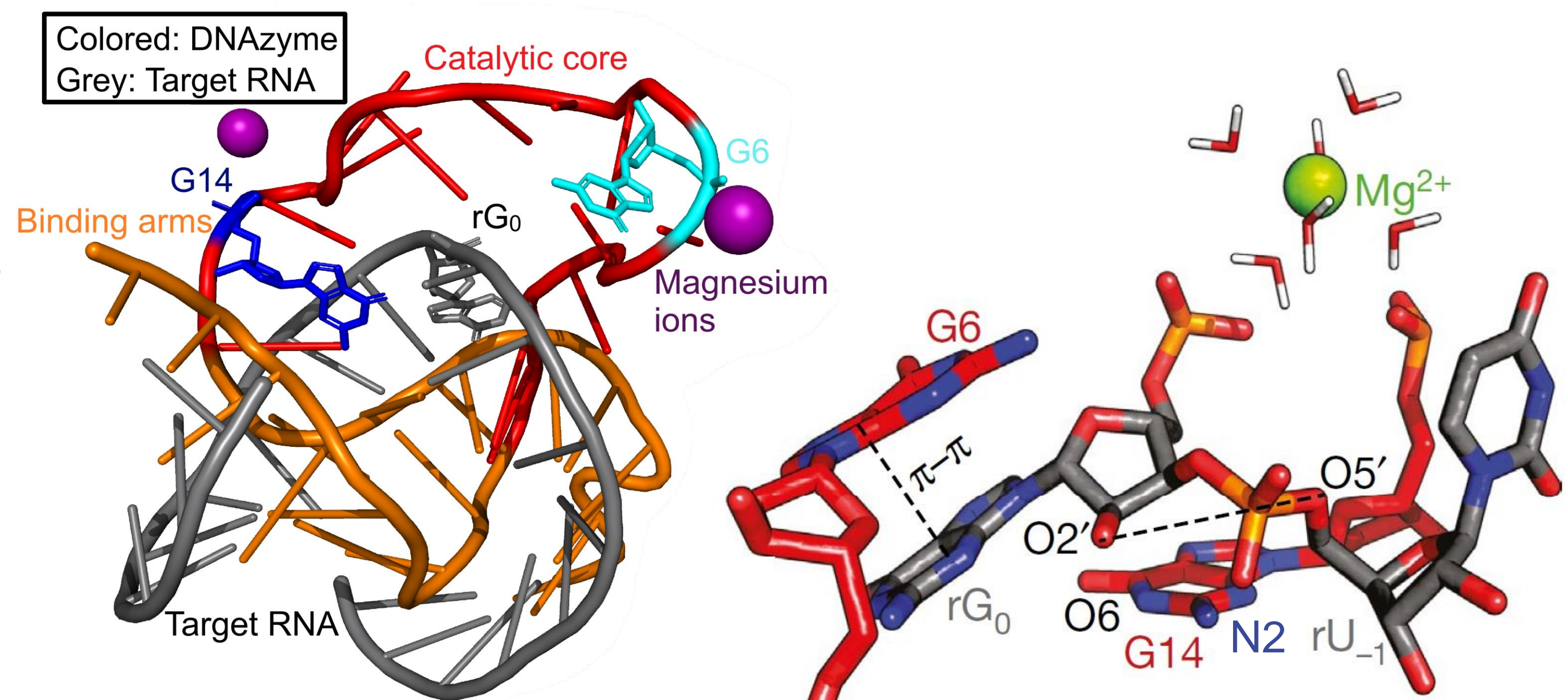
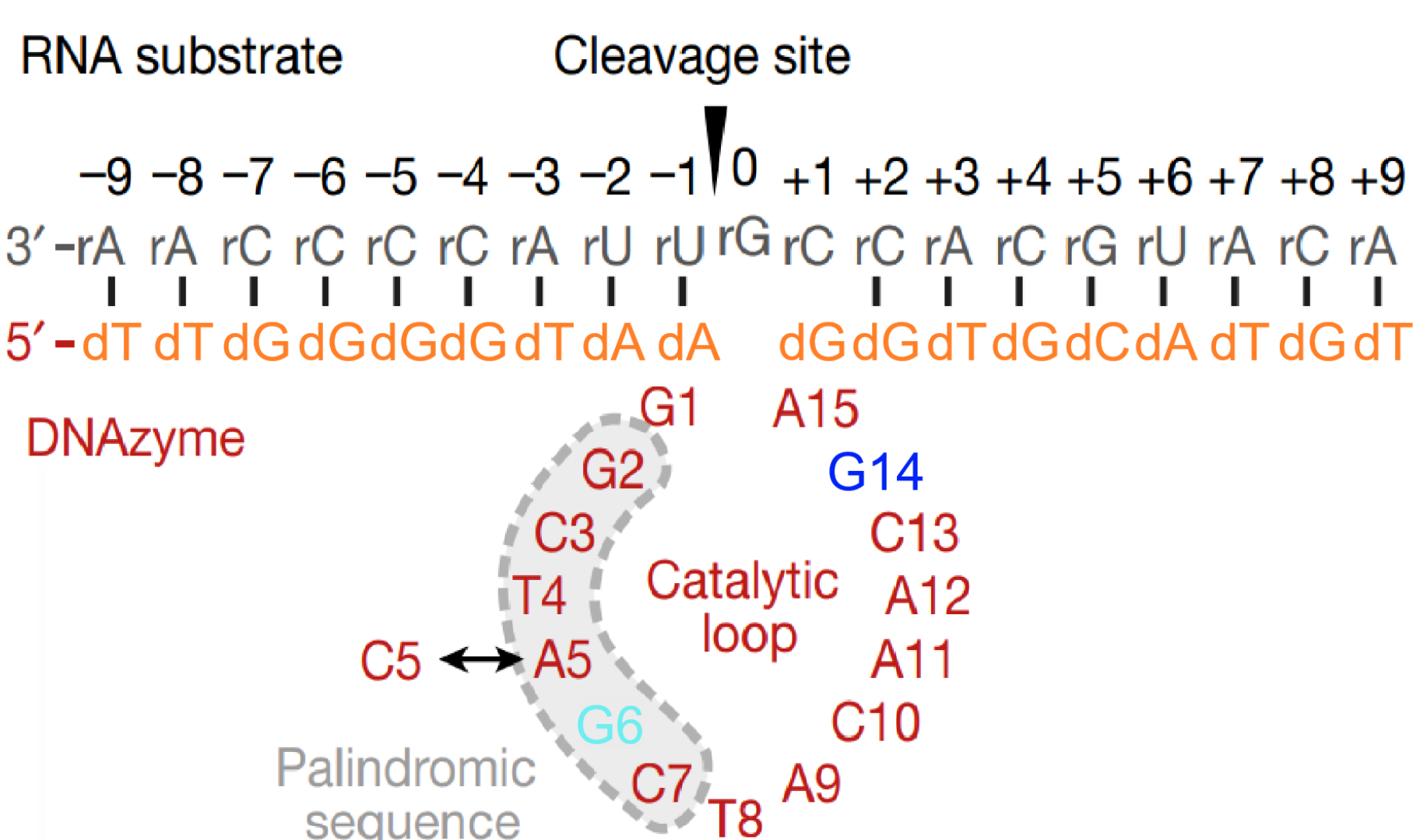
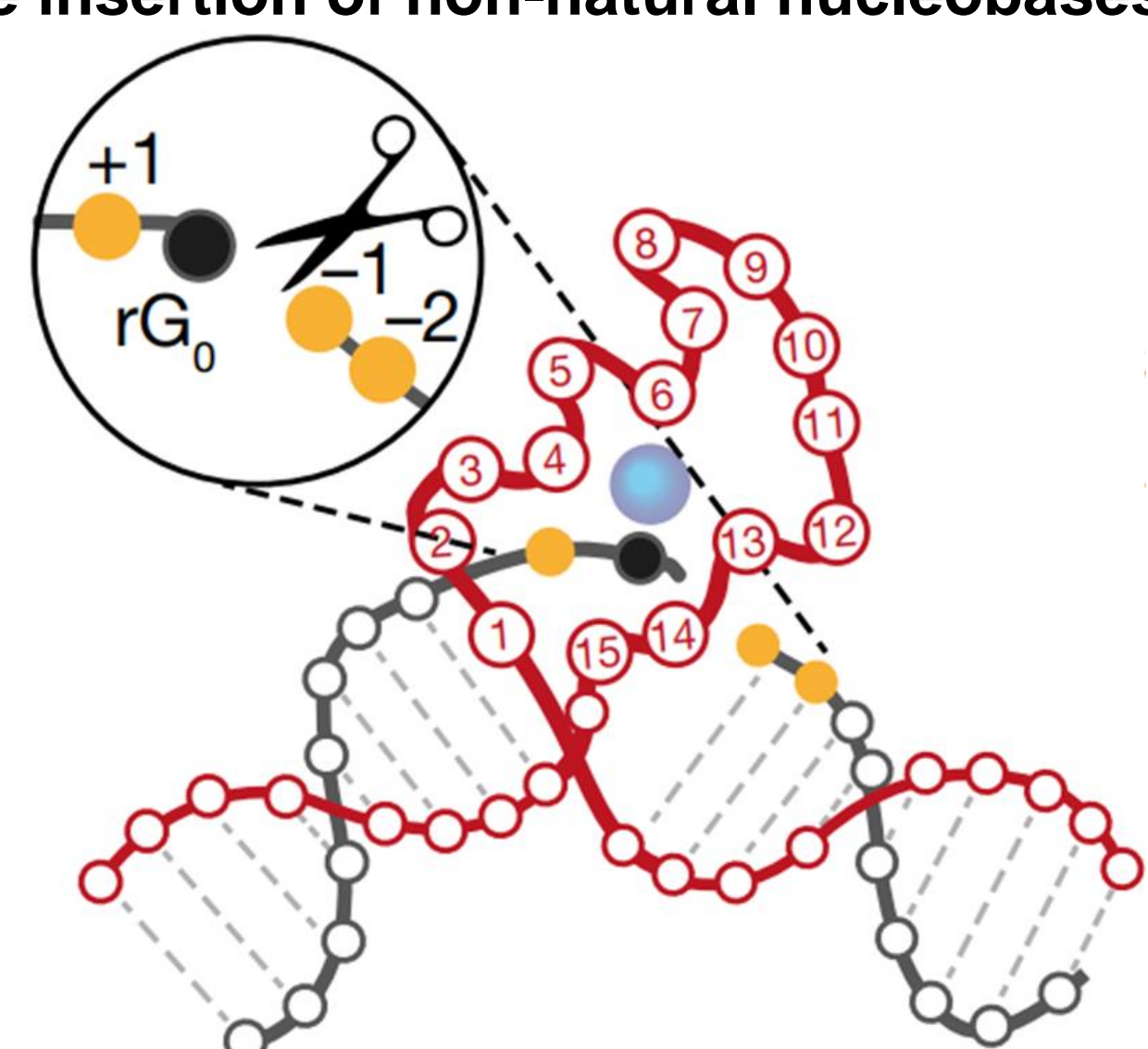
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## Introduction

The DNA is most known as the blueprint of protein synthesis, being the very first step into our life. Recently, a new function and the mechanism of **DNA** strands with catalytic activity, that cleave RNA, dubbed **DNAzymes**, has been revealed.<sup>[1]</sup> By molecular dynamics (MD) simulations in connection with experimental validation, we are investigating **how this cleavage conformation, the in-line attack (ILA), is formed and how non-natural nucleobases can be used to stabilize it**. This question is addressed on **DNAzyme 10-23**, a **DNAzyme with a wide range of RNA targets, able to cleave them with high selectivity**.<sup>[2]</sup> To increase the *in vivo* activity, we investigate the effect of non-natural nucleobases on the ILA formation, the molecular prerequisites of the ILA formation, the  $Mg^{2+}$  binding, and the  $\pi\pi$ -stacking of involved bases. In combination with experimental validations, we provide **insights into the cleavage mechanism of DNAzyme 10-23, improvements made by the insertion of non-natural nucleobases, and predictions for further modifications**.



## Distances and angles involved in in-Line attack formation

The **angle** between  $O2'$  of  $rG_0$  of the RNA and both the phosphate and an oxygen of  $rU_{-1}$  must be  $>120^\circ$  while the **distance** between the  $O2'$  of  $rG_0$  and  $N2$  of  $G14$  is less than  $4 \text{ \AA}$ .

**In-line attack needs both requirements met (blue box).**

**Both G14 and G6 are involved in the ILA formation.** 4 modifications exchanging those were run (CP1-CP4), while an **unmodified DNAzyme 10-23 (WT)** and a thioguanine at G14 (**TG**) were used as references with previous works, yielding similar results.<sup>[1]</sup> **Only in WT and CP4 an ILA formation was observed.**

Wildtype (WT)

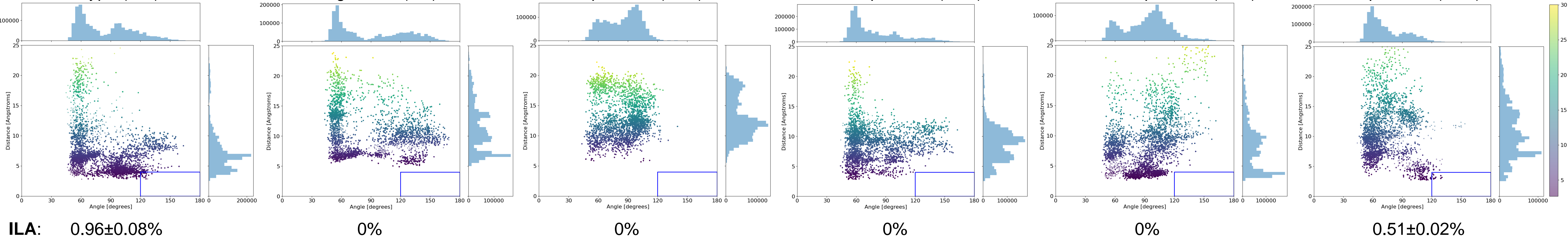
Thioguanine (TG)

Compound 1 (CP1)

Compound 2 (CP2)

Compound 3 (CP3)

Compound 4 (CP4)



ILA: 0.96±0.08%

0%

0%

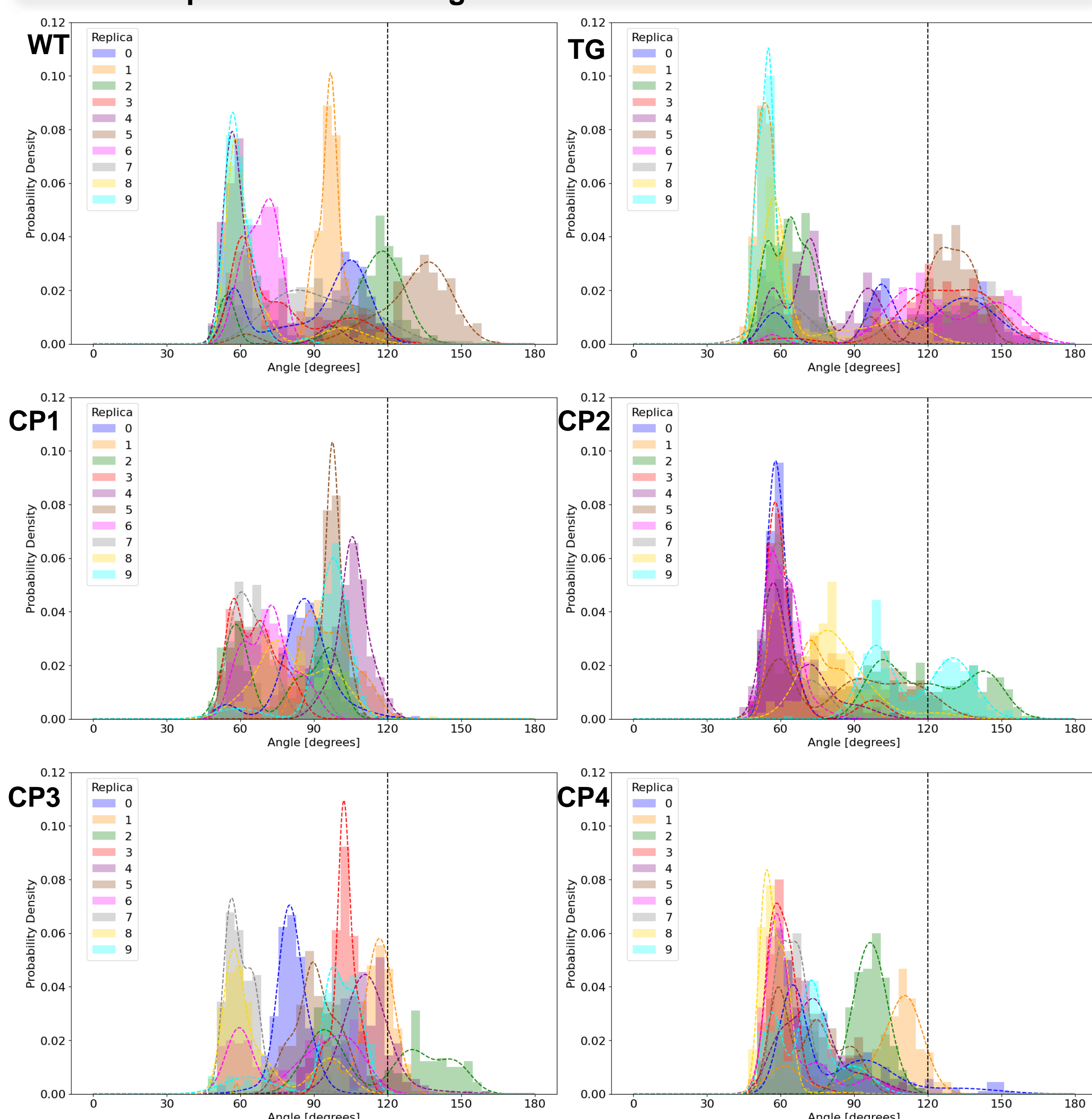
0%

0%

0.51±0.02%

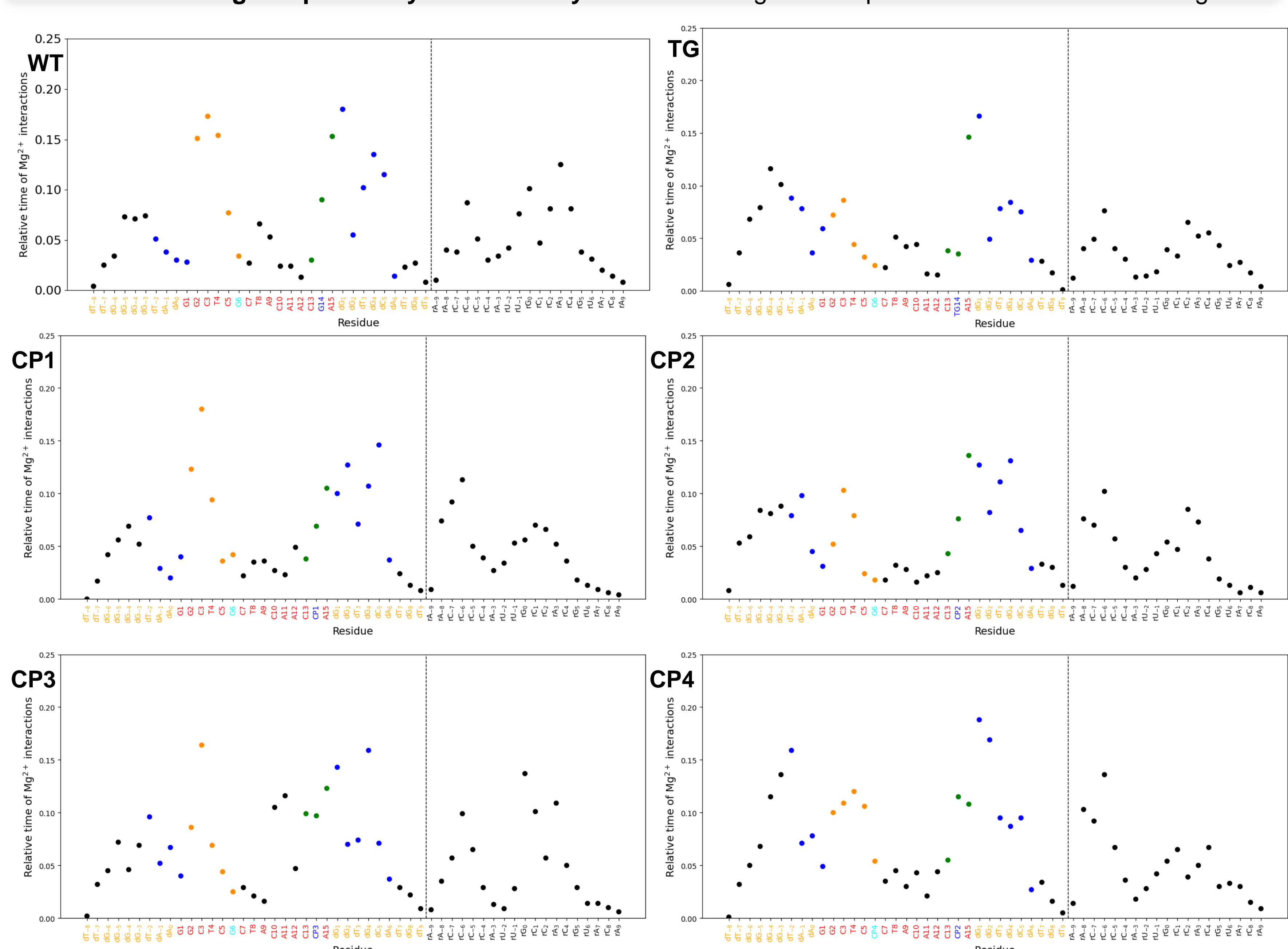
## Potential attack angle distribution

The **distribution of attack angles** was investigated to see how any modification changed them. The experimentally **most active modification (TG)** showed the **most potential attack angles** but lacked the distance to initiate ILA.



## Magnesium ion distribution in the DNAzyme

The  $Mg^{2+}$  binding was investigated, were the **metal binding sites of the unmodified DNAzyme** have been colored blue, orange and green. A **new metal binding site** is observed in modification **CP4**, potentially leading to a **reduction of  $Mg^{2+}$  dependency of the DNAzyme** due to being more exposed to the solvent in this region.



## Summary

Unbiased MD show **that the introduction of non-natural modification can significantly change the behavior of DNAzymes**. While the unmodified DNAzyme 10-23 (WT) showed the highest ILA formation, the **experimentally 6-times more active thioguanine modification (TG)** showed **no ILA**. This is in line with previous simulations, were also no ILA formation was detected for thioguanine.<sup>[1]</sup> Both the In-line attack formation and  $Mg^{2+}$  binding change upon the insertion of non-natural nucleobases, aiding in **decreasing their strong cofactor dependency**.<sup>[3]</sup> Given the current simulations, the greatest challenge is closing the distance between  $O2'$  and  $N2$  to initiate the ILA formation. Most modification can reach proper attack angles and even form an ILA but simply can't close the distance. **Experimental verification showed that beneficial effects observed in the formation of the ILA also stabilize the ground state**, increasing the energy barrier to cleave the RNA. To overcome this energy barrier, further modifications of the involved atoms may be necessary.

## References

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